

Chemistry If8766 Answers Ph And Poh

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pH and pOH The pH of a solution indicates how acidic or basic that solution is. pH range of 0-7 acidic 7 neutral 7-14 basic Since ... [H+] pH [OH-] pOH Acidic or Basic 1. 10⁻⁵ M 5 10⁻⁹ M 9 Acidic 2. 10⁻⁷ M 7 10⁻⁷ M 7 Neutral 3. 10⁻¹⁰ M 10 10⁻⁴ M 4 Basic 4. 10⁻² M 2 10⁻¹² M 12 Acidic 5.

pH and pOH Worksheet - Answer Key - SarahChem

HYDROLYSIS OF SALTS Salt solutions may be acidic, basic, or neutral, depending on the original acid and base that formed the salt. Strong acid + strong base neutral salt

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The Arrhenius definition of acid and base is limited to aqueous (that is, water) solutions. Although this is useful because water is a common solvent, it is limited to the relationship between the H⁺ ion and the OH⁻ ion. What would be useful is a more general definition that would be more applicable to other chemical reactions and, importantly, independent of H₂O.

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