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Favourite answer. 1. balance everything except H and O by counting symbols on both sides: there are 2 C atoms on the left and only one on the right so insert a 2 like this:  $C_2H_5OH + O_2 \implies 2CO_2 + \dots$

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Favorite Answer. 1... Tricarbon octahydride is a very fancy name for Propane.  $C_3H_8$ . The equation otherwise is correctly balanced.  $C_3H_8 + 5O_2 = 3CO_2 + 4H_2O$  (or  $4HOH$ ). Your teacher was trying to trick...

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#### HOW TO COMPLETE WORD EQUATIONS IN CHEMISTRY? | YAHOO ANSWERS

Reaction information is shown using word and symbol equations. The simplest ratio of atoms of each element in a compound is called the empirical formula. Mass is conserved in chemical reactions.

#### CHEMISTRY WORD EQUATIONS? | YAHOO ANSWERS

This is the most common type of chemical equation. In this chemical equation, a new product is formed by combining two to three combinations of reactants. For instance,  $H_2 + O_2 \rightarrow 2H_2O$ . This is a chemical equation where two atoms of hydrogen are combined to form a product, water. This is why this reaction is called as synthesis reaction.

#### CHEMISTRY WORD EQUATIONS QUESTIONS ANSWERS

A chemical equation tells you which chemicals reacted together (the reactants) and the new chemicals that were made in the reaction (the products). An example is the reaction of sodium with oxygen: sodium + oxygen  $\rightarrow$  sodium oxide. Sodium and oxygen are the reactants and there is just one product, sodium oxide. Read More.

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Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead.  $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$  Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas.  $2AlBr_3 + 3Cl_2 \rightarrow 2AlCl_3 + 3Br_2$  Single Replacement 3.

*Answered: Show the word equation, chemical...* | bartleby

Chemical reactions have the same number of atoms before the reaction as after the reaction. Balancing chemical equations is a basic skill in chemistry and testing yourself helps retain important information. This collection of ten chemistry test questions will give you practice in how to balance chemical reactions.

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\*Response times vary by subject and question complexity. Median response time is 34 minutes and may be longer for new subjects. Q: Calculate the mole fraction of each solute and solvent ; a. 583 g of  $H_2SO_4$  in 1.50 kg of water - the ... Q: 15.54 Draw condensed formulas for the carboxylic acid and ...

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$MgO + HNO_3 \rightarrow Mg(NO_3)_2 + H_2O$ . OXYGEN: There is 1 Oxygen Atom on the left and 1 on the right so there are equal number of Oxygen Atoms on both sides so is kept the same (remember that you are counting the Nitrate Group as separate group, so do not count the Oxygen Atoms in this Group)  $MgO + HNO_3 \rightarrow Mg(NO_3)_2 + H_2O$ .

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